



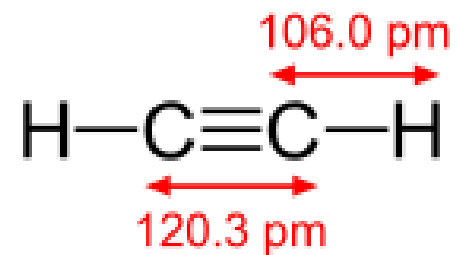
Alkyny



<https://www.wikihow.com/Set-Up-an-Oxy-Acetylene-Torch>



<https://jtcylinderracks.com/common-uses-for-acetylene-cylinders/>



<https://en.wikipedia.org/wiki/Acetylene>



<https://fractory.com/gas-welding-explained/>

Alkyny - klasifikace



ethyn
(acetylen)

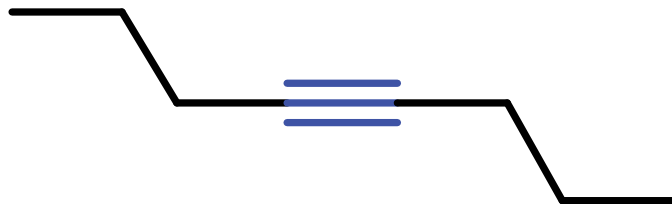


terminální alkyn



interní alkyn

Alkyny – názvosloví (přípona **-yn**)

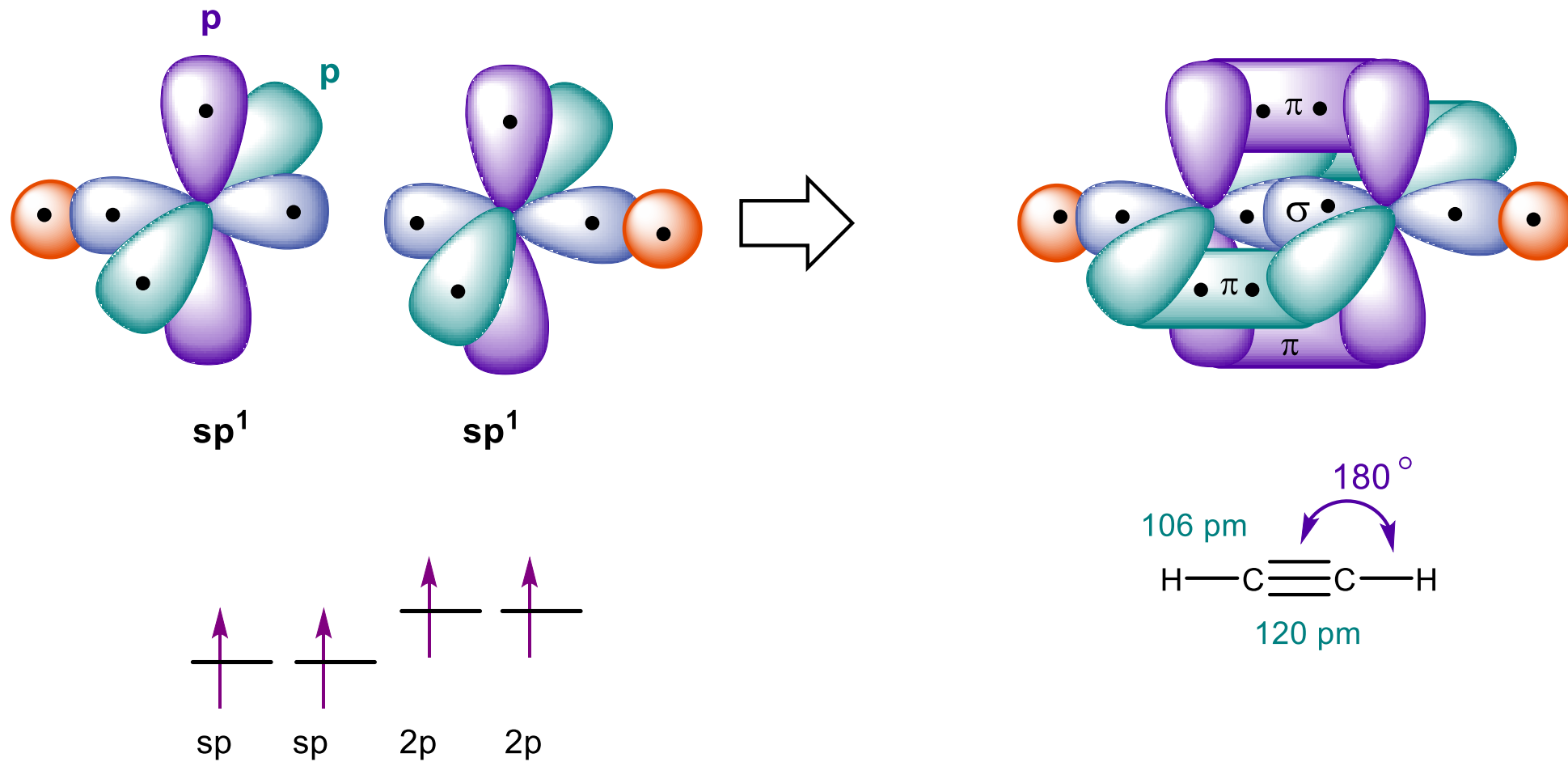


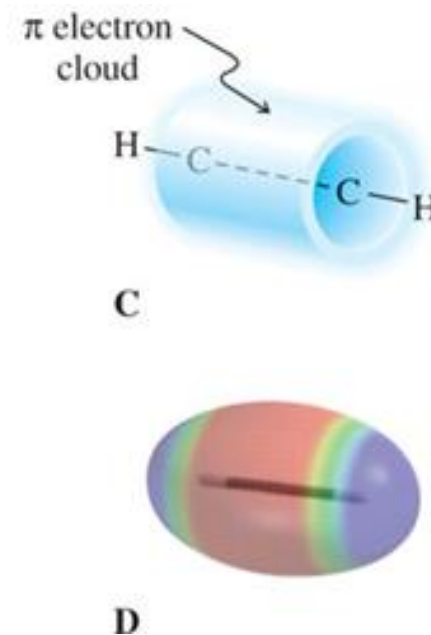
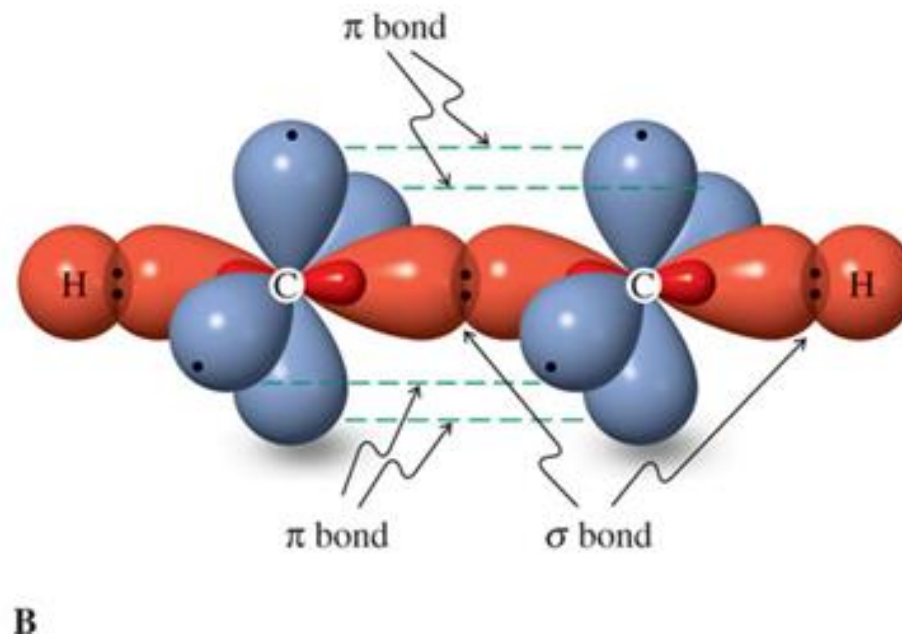
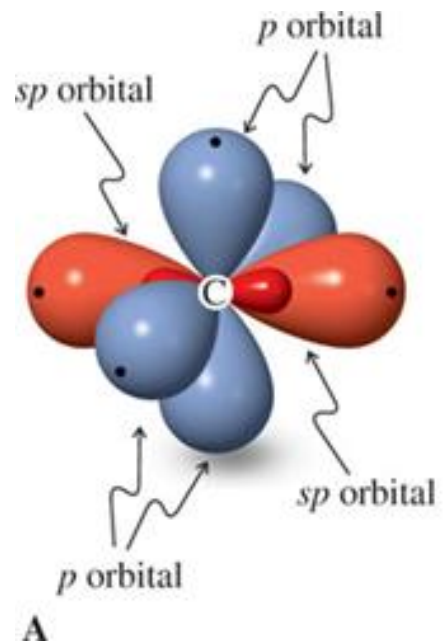
okt-4-**yn**



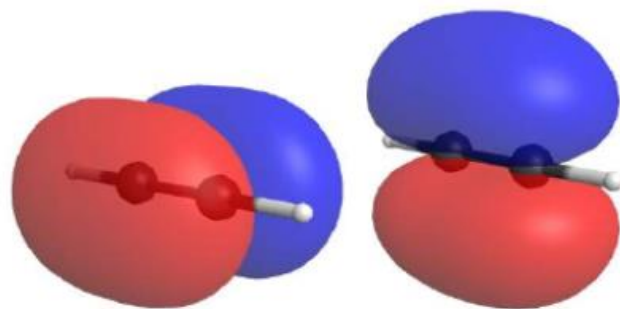
hexa-2,4-di**yn**

Molekulové orbitály – vznik trojné vazby $\text{C}\equiv\text{C}$

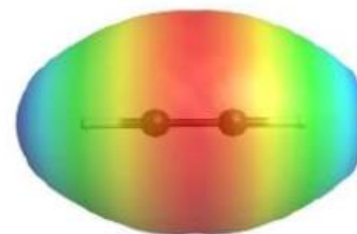




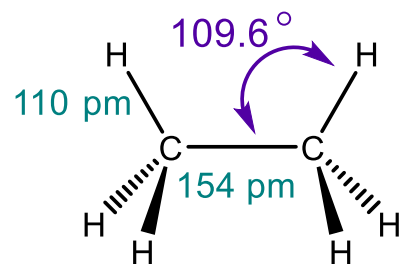
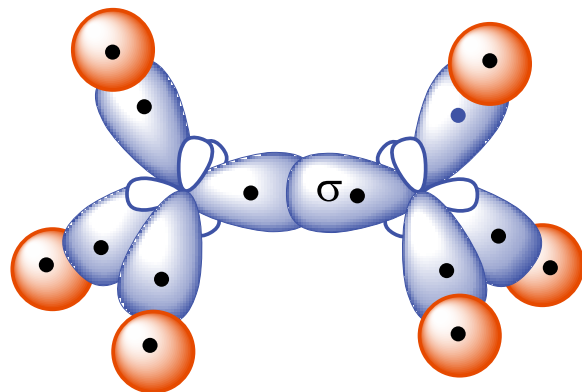
Example of the π bonds in ethyne



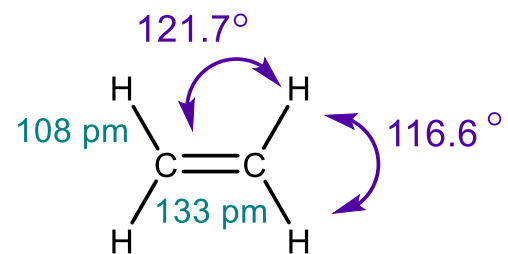
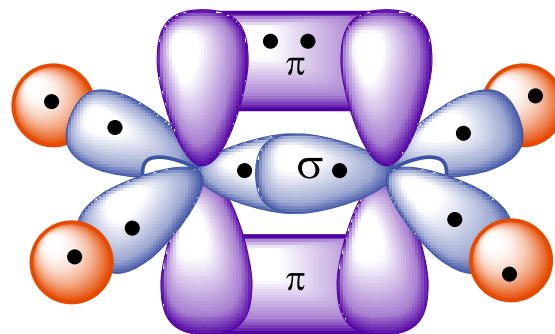
Electrostatic potential map of ethyne (red = high electron density and blue = low electron density)



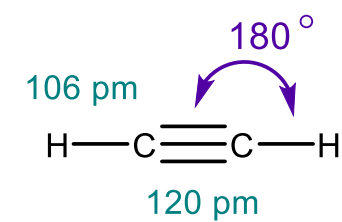
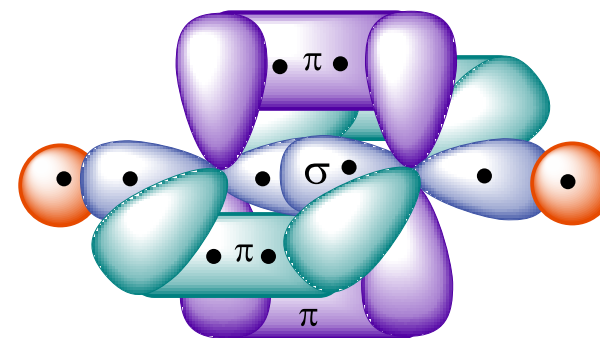
sp^3



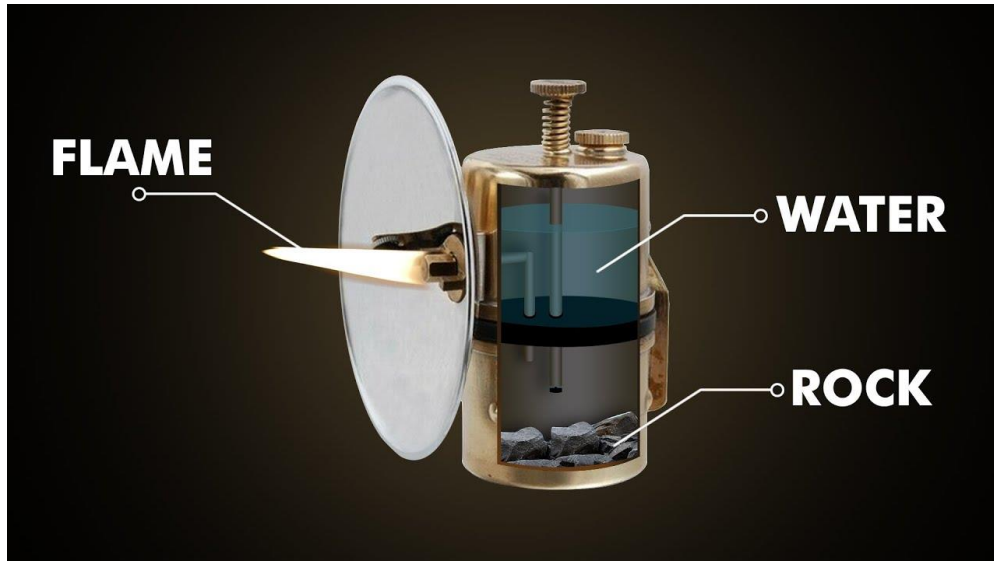
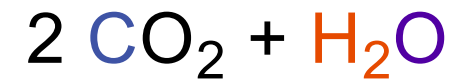
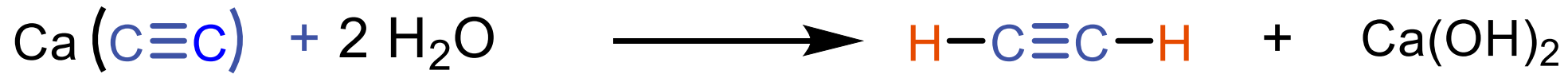
sp^2



sp



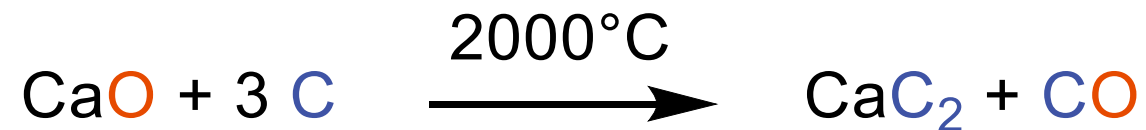
Karbid vápenatý – karbidová lampa



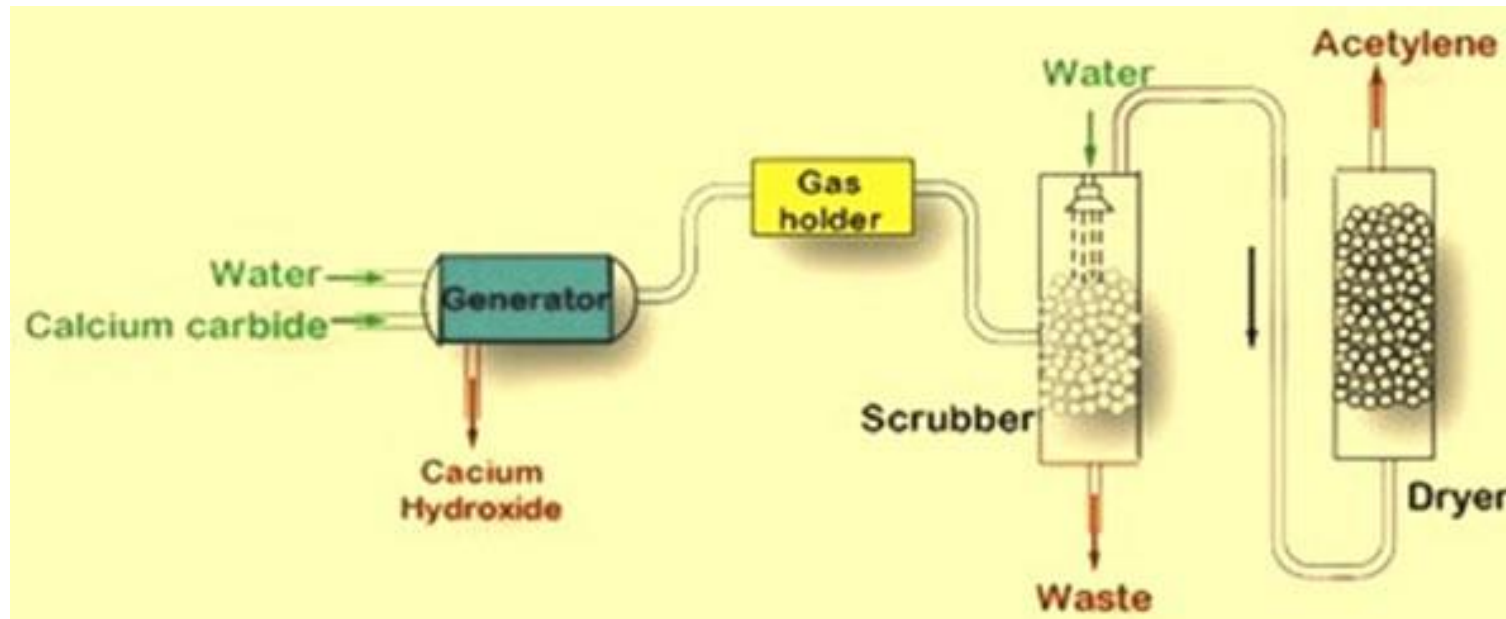
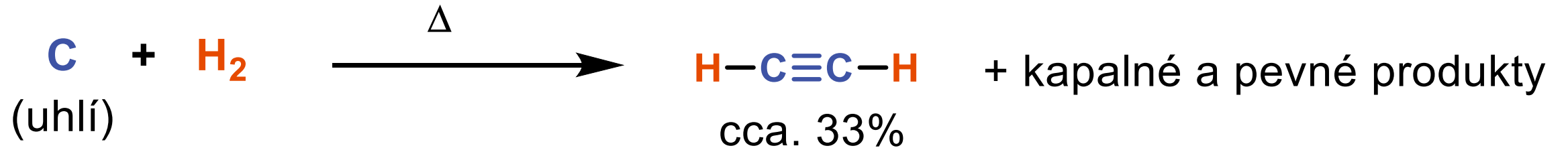
https://www.google.com/search?sca_esv=577421072&q=carbide+lamp&tbm=isch&source=lnms&sa=X&sqi=2&ved=2ahUKEwiAh5zPw5iCAxWUgf0HHaVIADIQ0pQJegQIDRAB&biw=1280&bih=563&dpr=1.5#imgrc=HbqIB_RsgffMJM



<https://camachem.com/en/calcium-carbide.html>

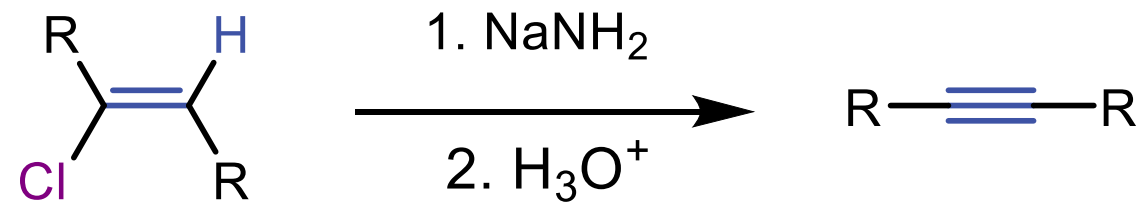
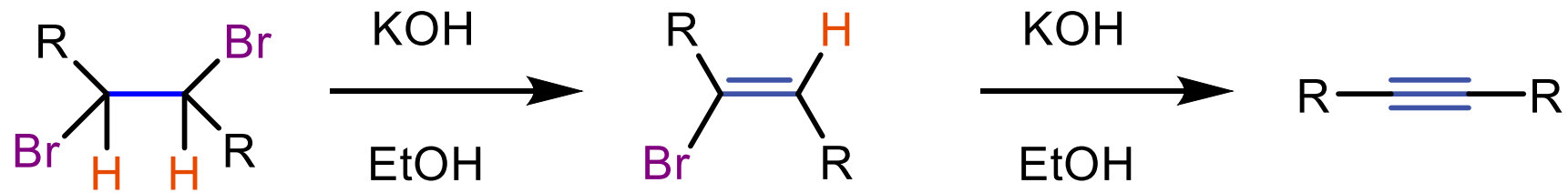


Průmyslová výroba acetylenu

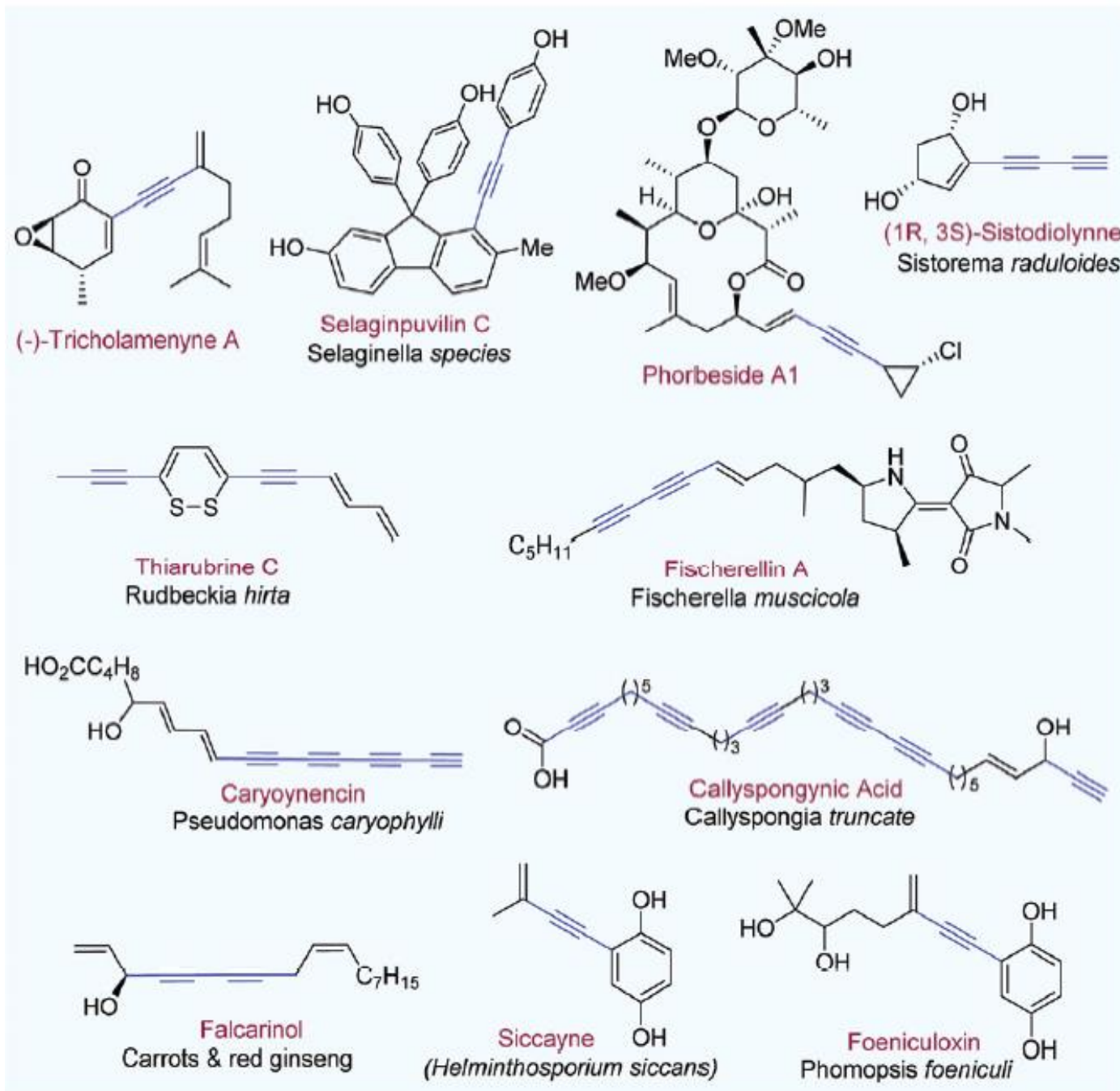


Příprava alkynů – eliminační reakce

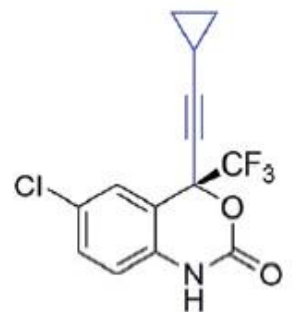
Eliminace halogenovodíků s alkylhalogenidů a alkenylhalogenidů - silnou bází



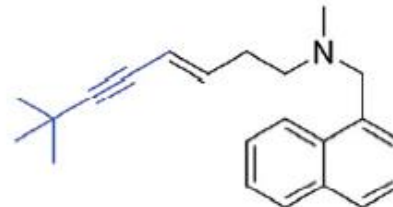
Motivy alkyňů v přírodních látkách



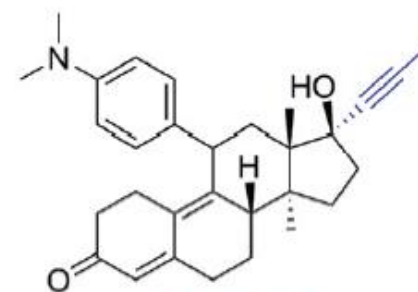
Motivy alkynů v léčivech



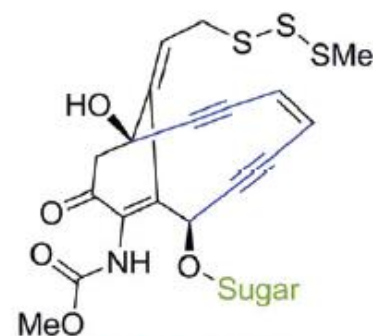
Efavirenz
Treats HIV/AIDS



Terbinafine
Antifungal drug



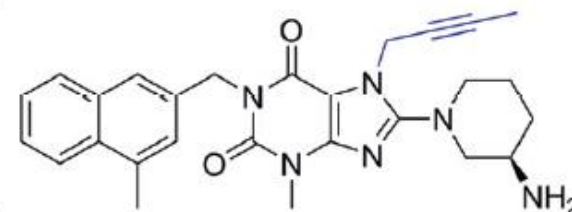
Mifepristone
For abortion



Calicheamicin
Anticancer



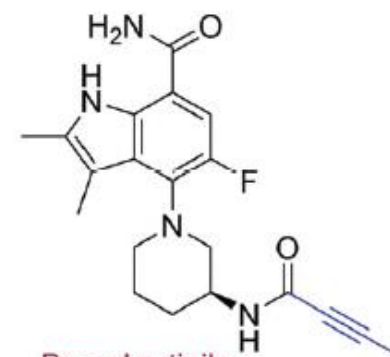
Mavoglurant
(MGLuR5) Antagonist



Linagliptin
Anti-diabetic



Tazarotene
treats Acne vulgaris
and Psoriasis



Branebrutinib
BTK inhibitor

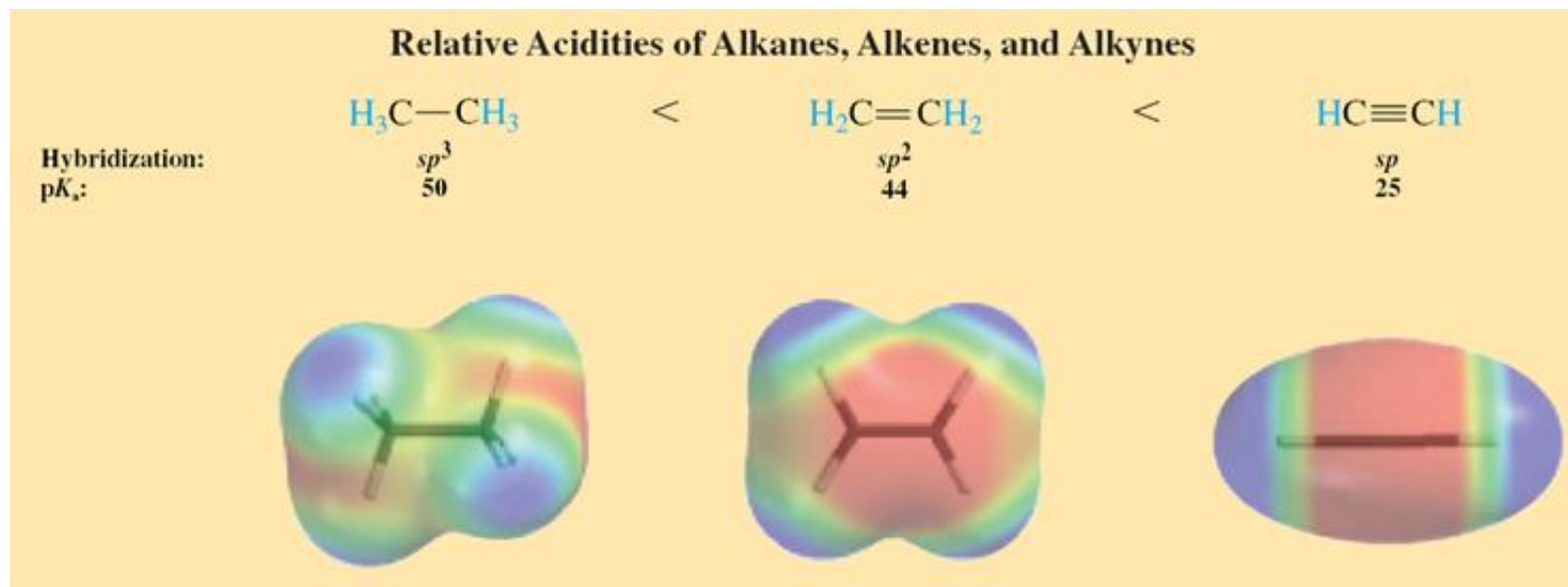
Reaktivita alkynů

Reakce trojné vazby
(adice)



Reakce koncové C-H
(deprotonace, alkylace.....)

Terminální alkyne: Acidita

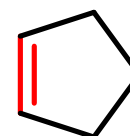
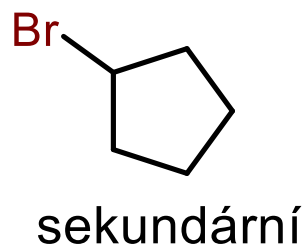
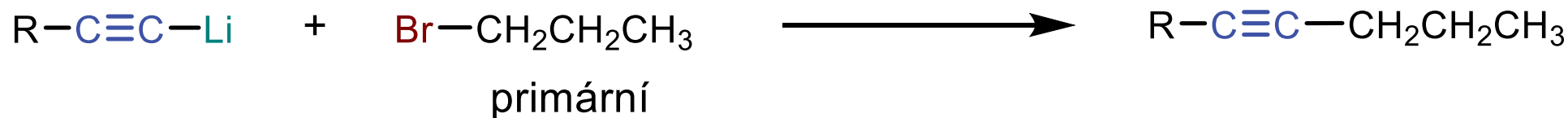


Deprotonace terminálních alkynů (a jejich reakce)

Deprotonace:

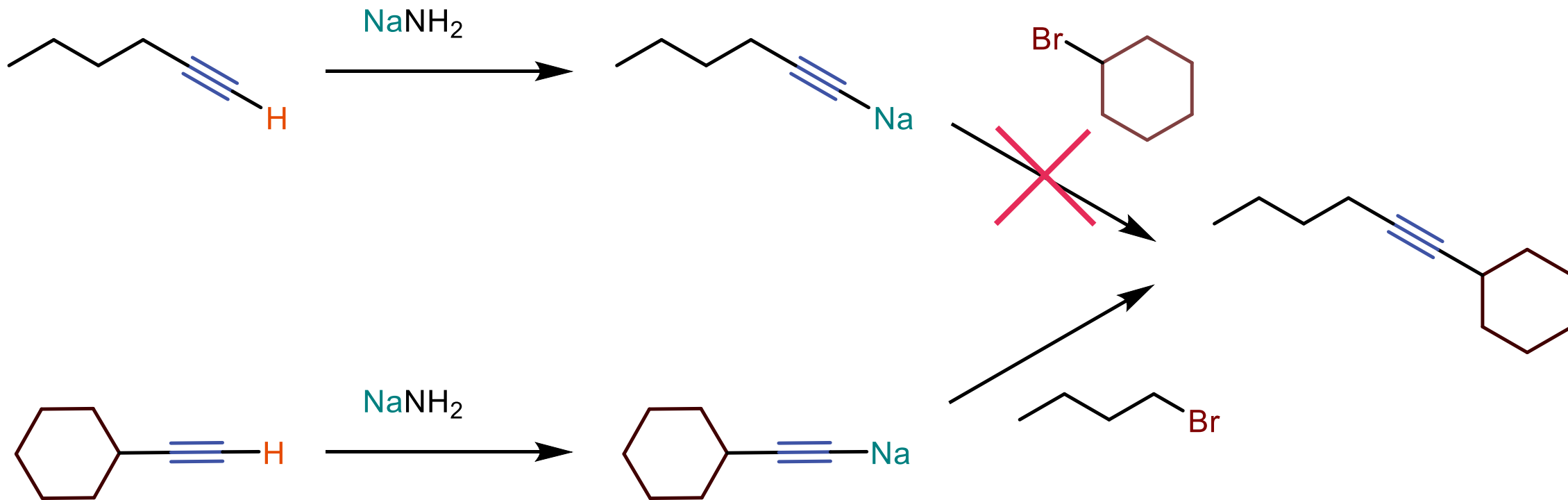


Alkylace:



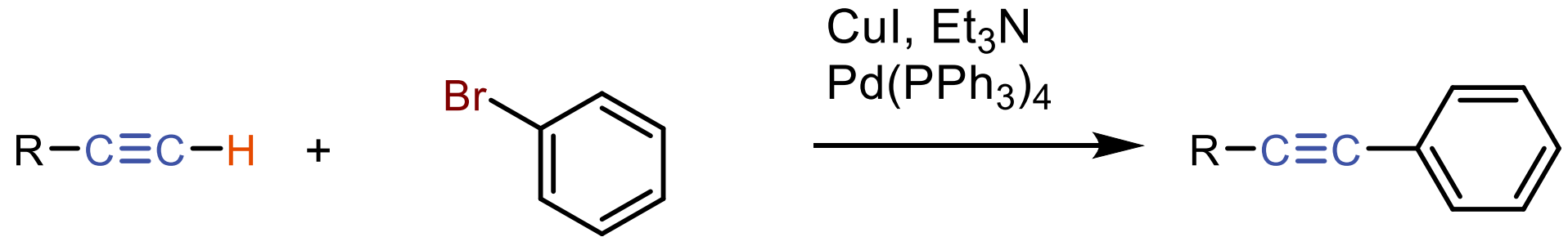
sekundární
a terciární alkylhalogenidy
podléhají eliminaci

Deprotonace + alkylace - organická syntéza

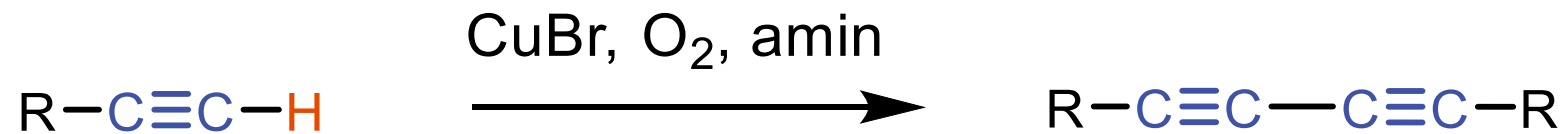


Další reakce terminálních alkynů

Sonogashirova reakce (coupling) – příprava aryl-alkynů

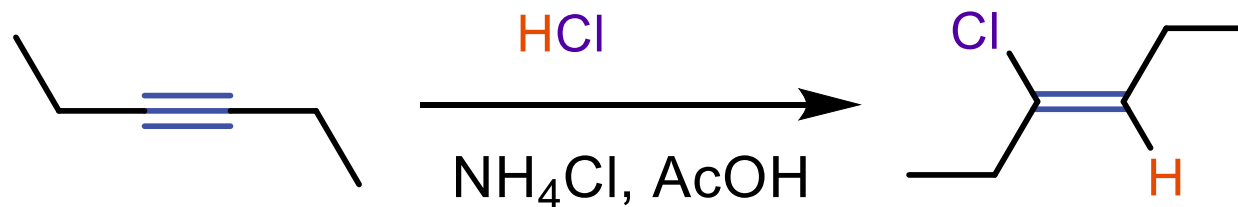
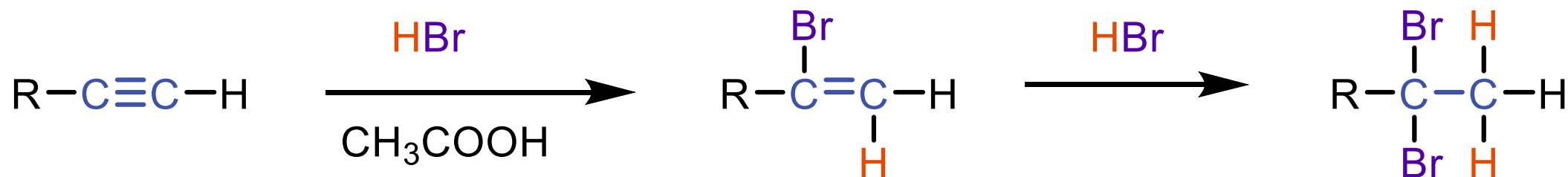


Glaserův coupling – příprava diacetylenů

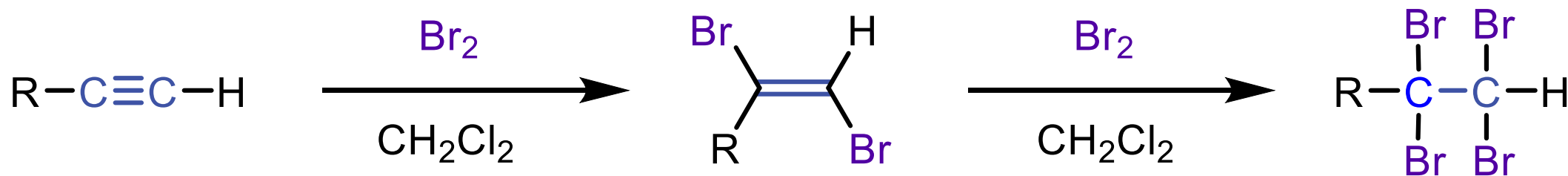


Adiční reakce na alkyny

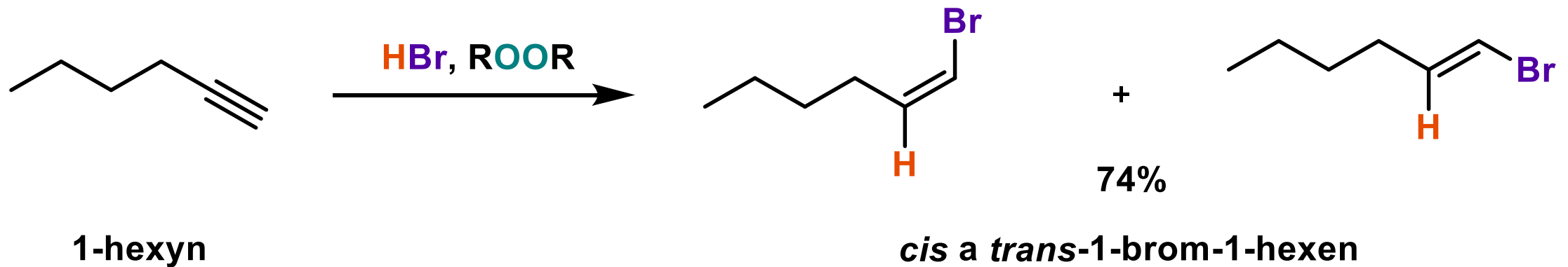
Adice halogenovodíků



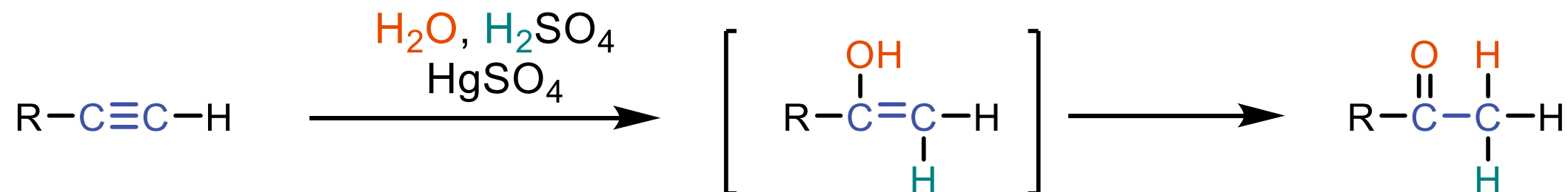
Adice halogenů



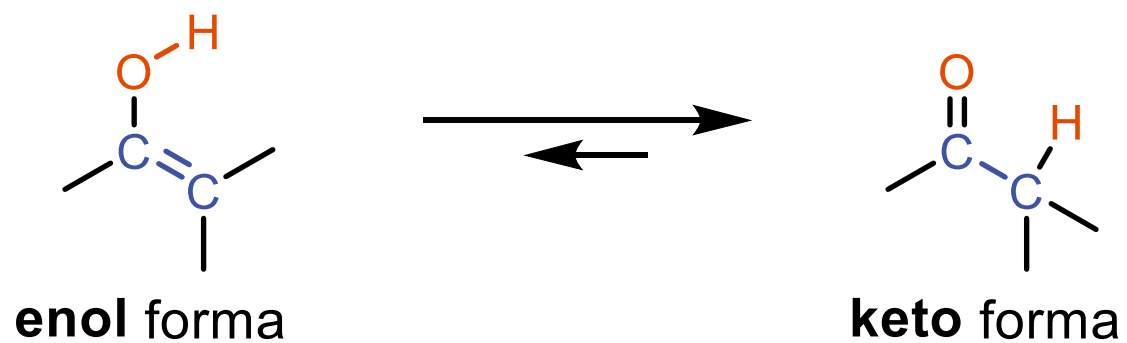
Radikálová adice HBr na alkyny – neprobíhá podle Markovnikova pravidla



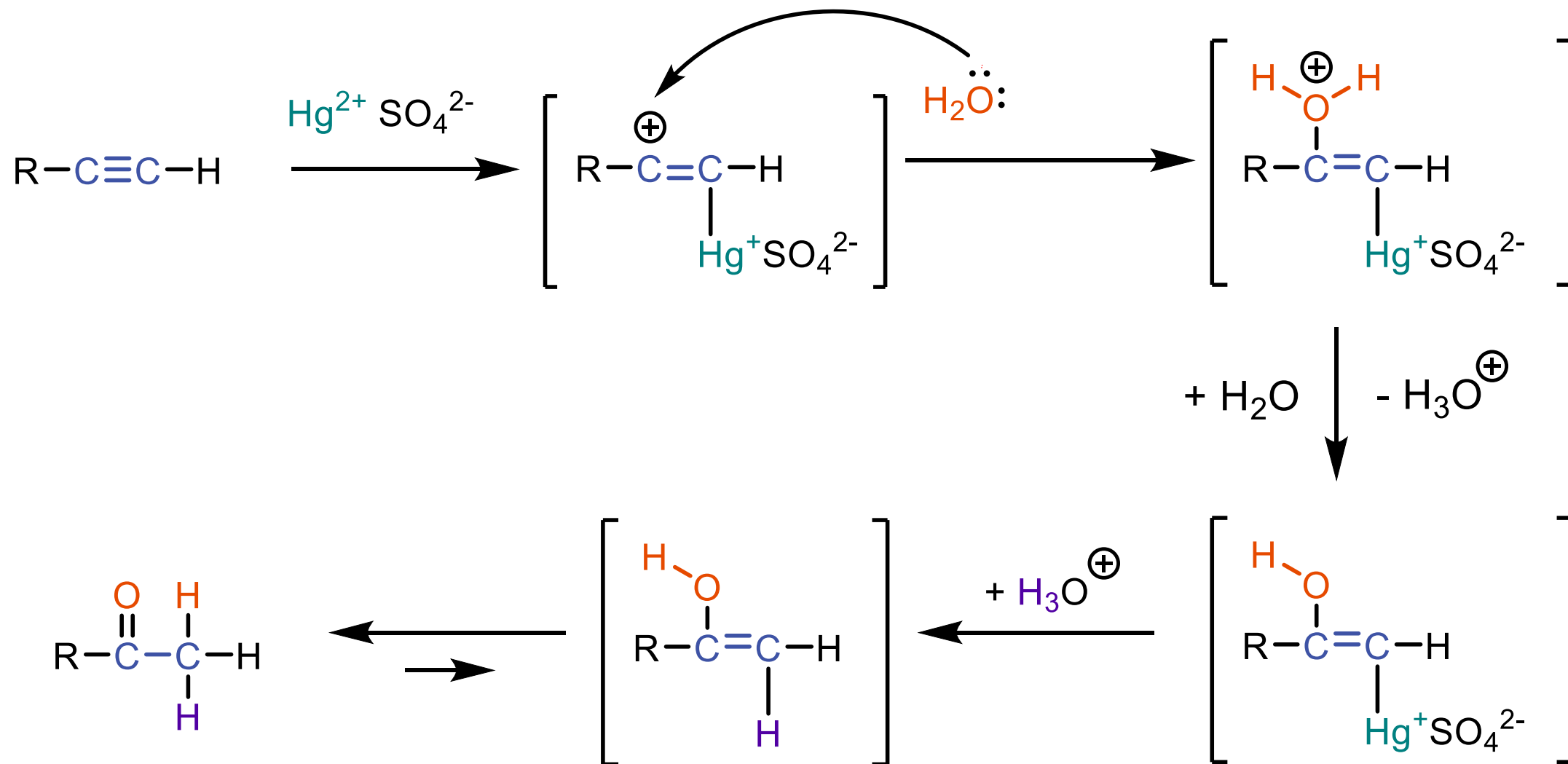
Hydratace alkynů katalyzovaná Hg^{2+}



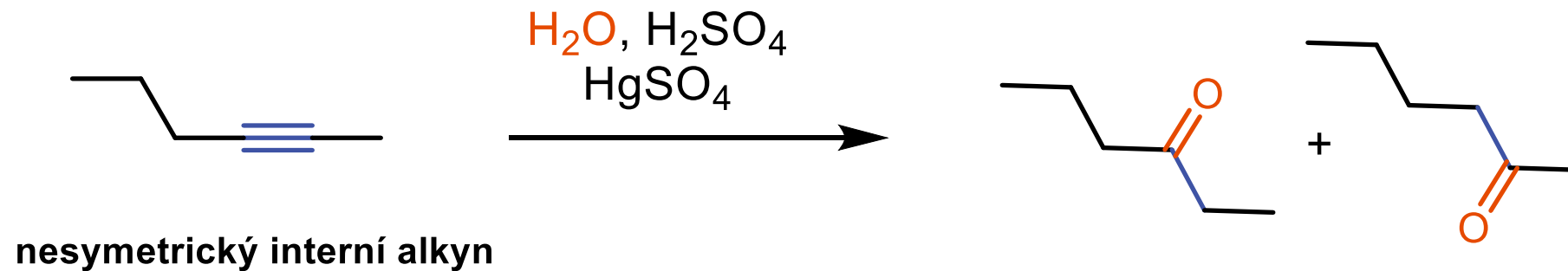
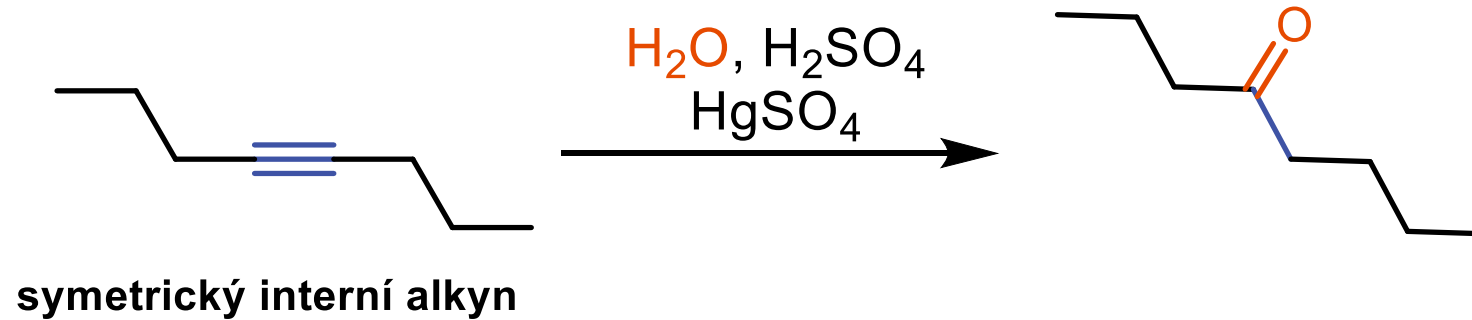
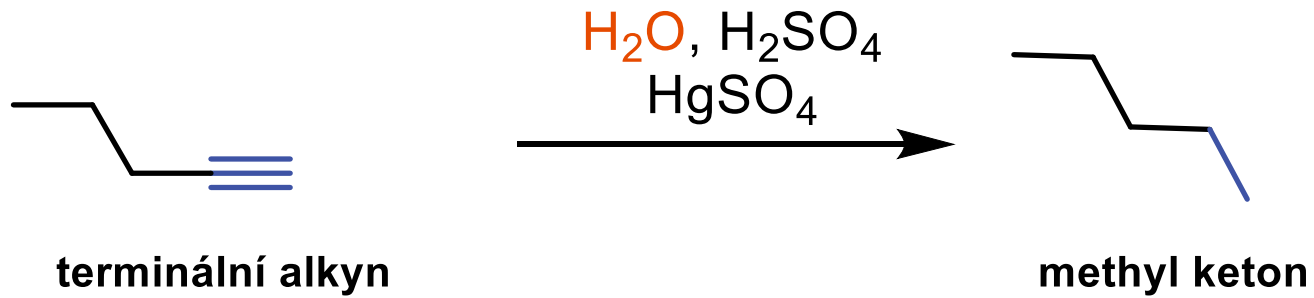
Keto-enol tautomerie



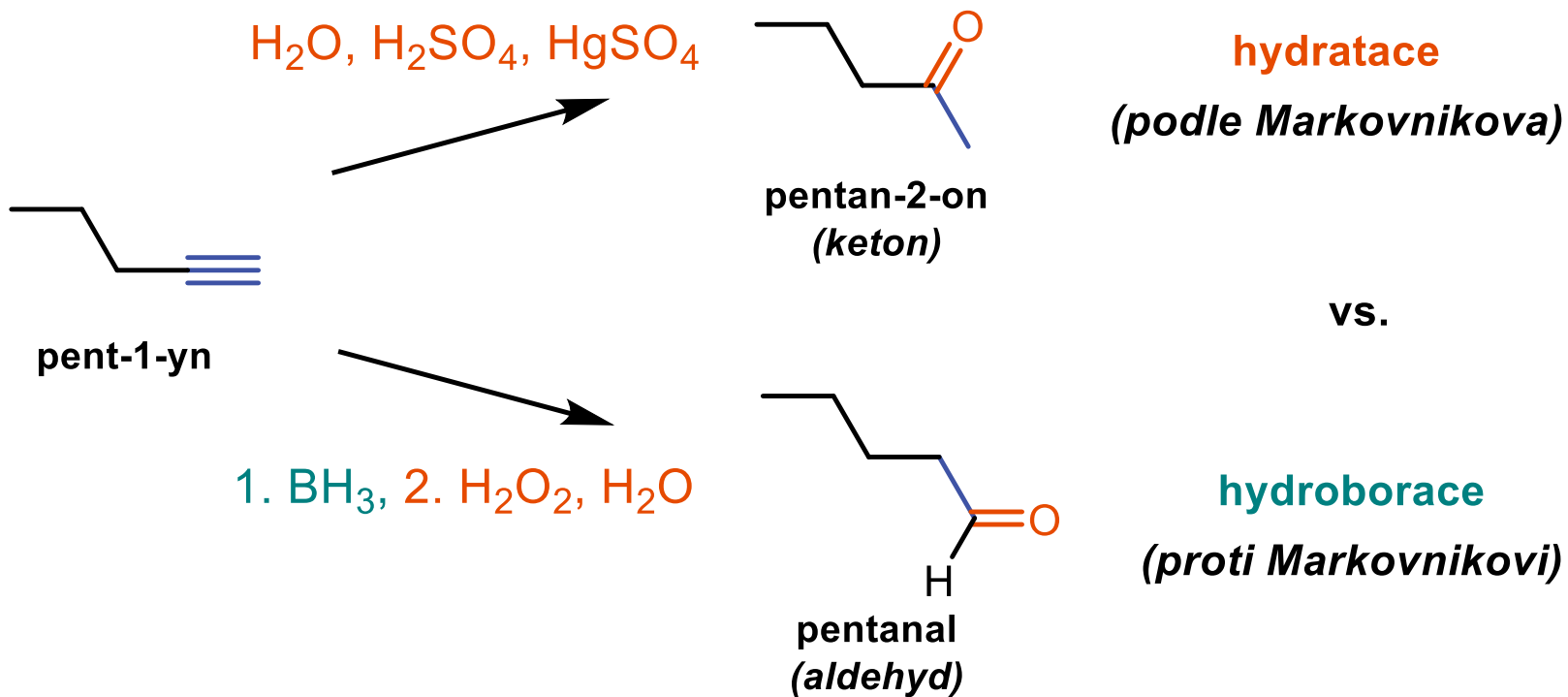
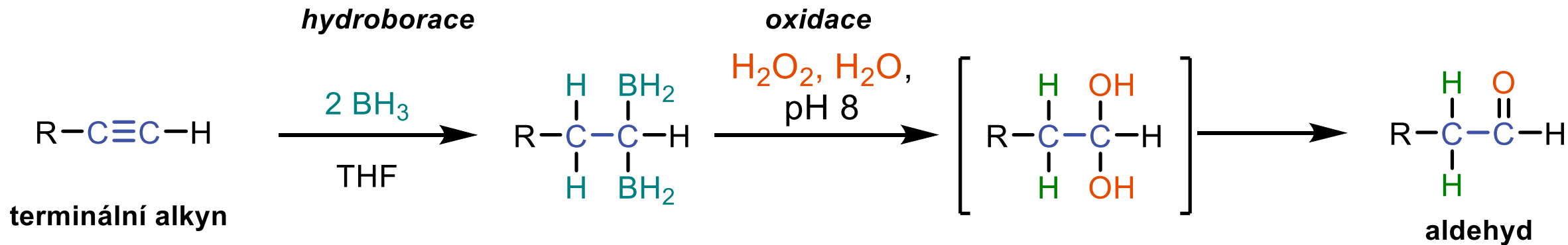
Mechanismus hydratace alkynů



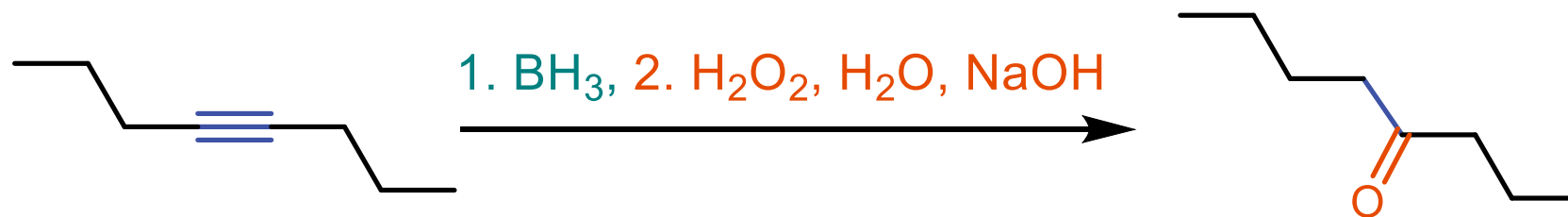
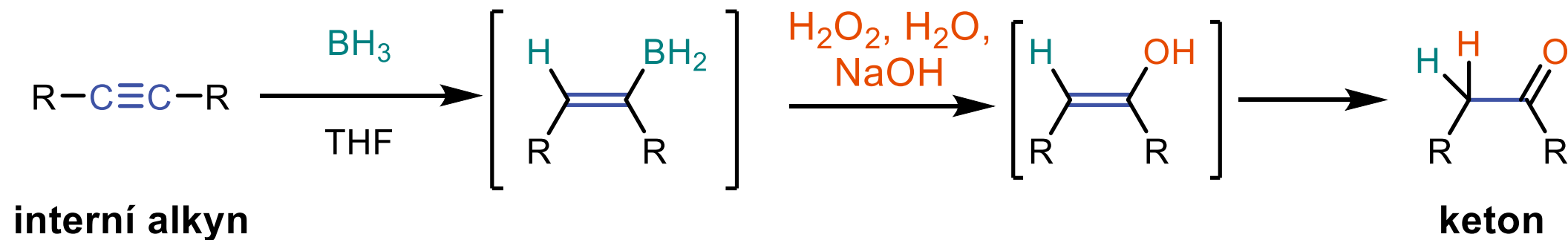
Hydratace alkynů – syntetické využití



Hydroborace alkynů

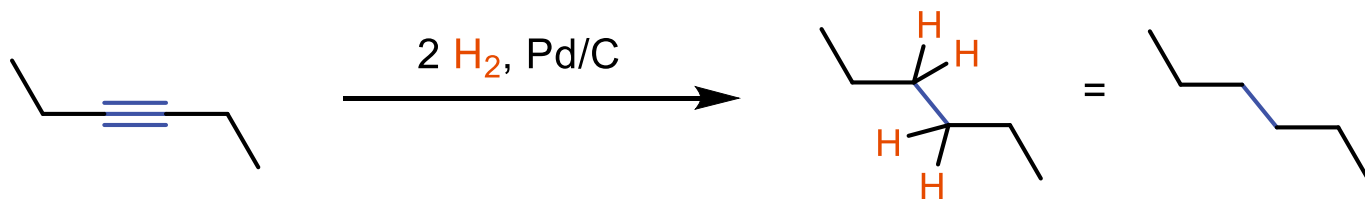


Hydroborace alkynů

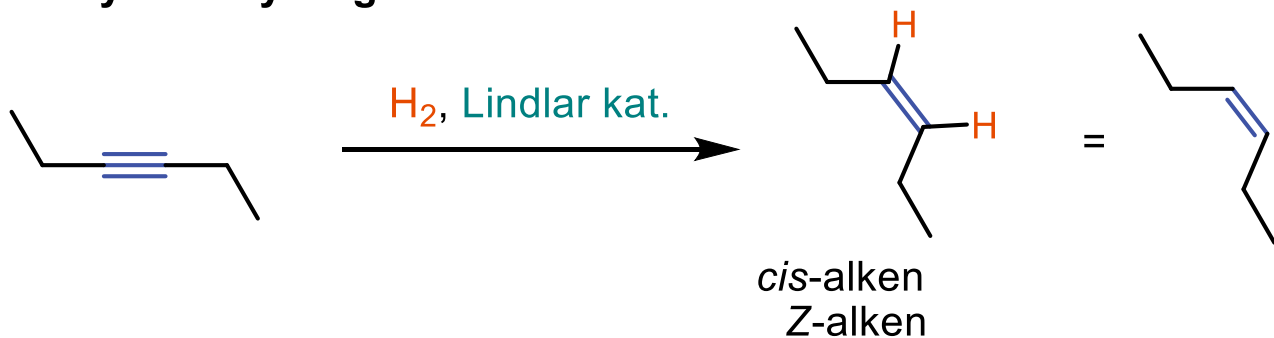


Redukce alkynů

Katalytická hydrogenace

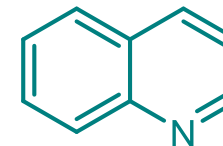


Katalytická hydrogenace - Lindlar

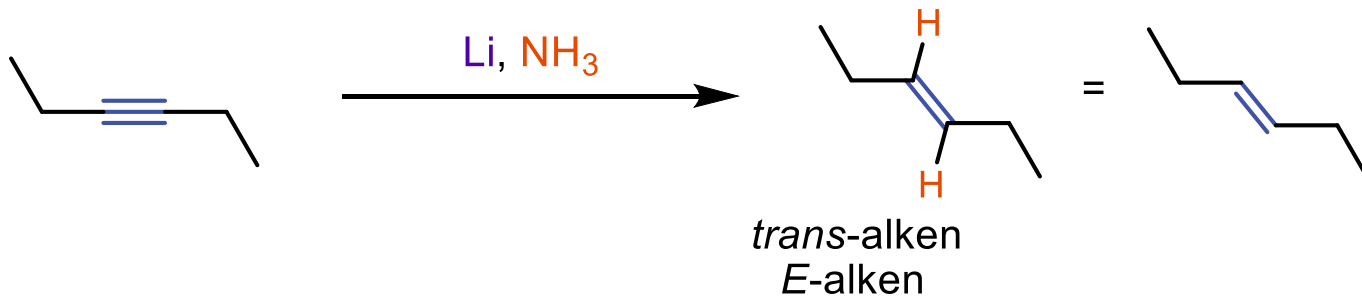


Lindlarův katalyzátor

5% Pd/CaCO_3 , Pb(OAc)_4 , chinolin



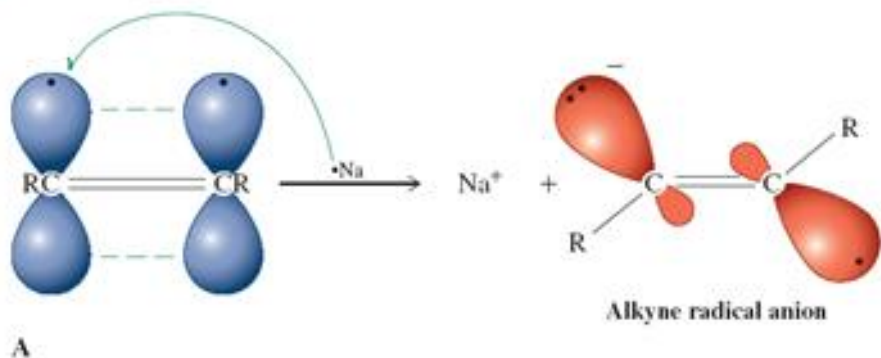
Redukce alkalickým kovem v kapalném amoniaku (Birchova redukce)



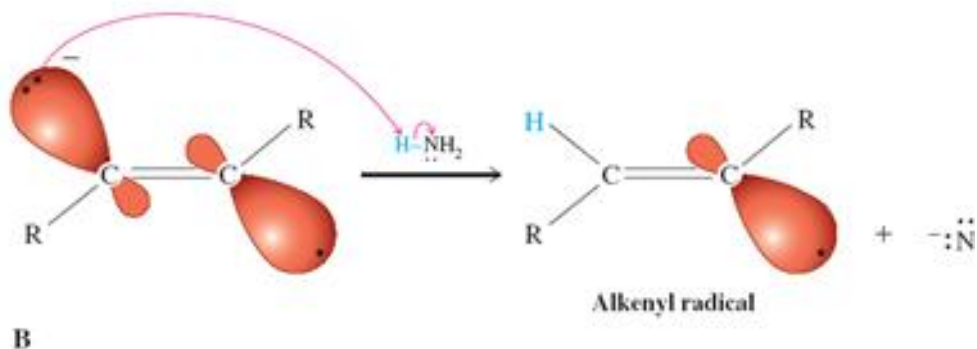
Birchova redukce - mechanismus

Mechanism of the Reduction of Alkynes by Sodium in Liquid Ammonia

STEP 1. One-electron transfer

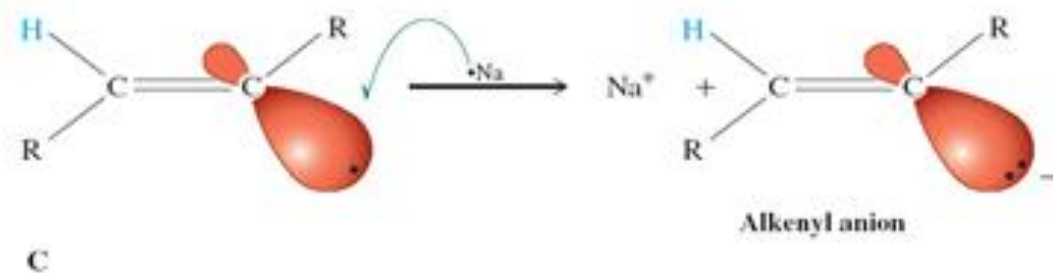


STEP 2. First protonation

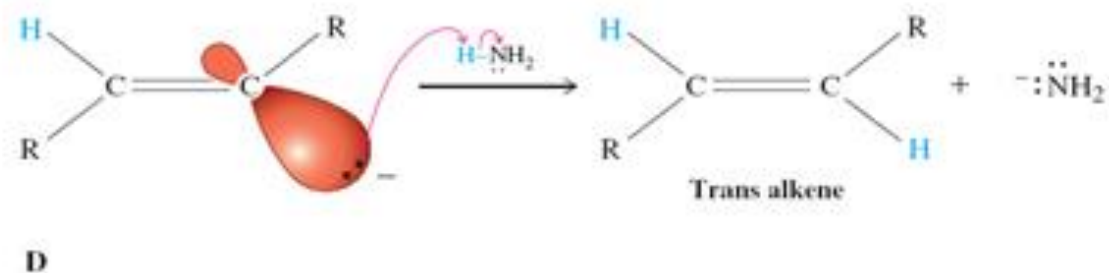


Druhý transfer elektronu je rychlejší než *cis/trans* izomerizace alkenylového radikálu.

STEP 3. Second one-electron transfer



STEP 4. Second protonation



Finální alken je stabilní vůči redukci.

Oxidativní štěpení alkynů

